

# 3-Hydroxymethyl-1,4-dihydro-4-oxoquinoline like compound with promising biological and complexing activity

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## Introduction

Hydroxymethyl group is present in many useful and biologically potent substances like sugars, kojic acid (Bajpai, 1982), ronicol (3-hydroxymethylpyridine) (Moncol, 2006; Broghammer, 1967) or hydroxymethylphenol – intermediate for bakelite (Baekeland, 1909).

On the other hand, 4-quinolones are a group of antibacterials with various biological activities if substituted in position 3- and they are anti-CNS and similar diseases agents if substituted in position 2- (kynurenic acid derivatives) (Turski, 2013; Szalardy, 2012; Fulop, 2009). Considering this, we were interested in the synthesis of 3-hydroxymethyl-1,4-dihydro-4-oxoquinoline.

4-Quinolones are a well-known group of chemotherapeutics with a broad spectrum of activities (Andriole, 2000; Milata, 2000) based on inhibition of the enzyme of bacterial topoisomerase II (DNA gyrase) and topoisomerase IV in Gram-positive species, thus inhibiting tertiary negative supercoiling of bacterial DNA (Gellert, 1976; Wang, 1985; Gootz, 1996) mostly bound to oxygen atoms in positions 3- and 4- of the fused pyrid-4-one-3-carboxylic acid moiety (Andriole, 2000). Therefore, it should be interesting to prepare and evaluate the corresponding structural motif, namely 3-hydroxymethylquinol-4-one.

Substructural search in database SciFinder<sup>®</sup> provided only various applications of formaldehyde with 4-quinolones or their precursors (Goldsworthy, 1982) but no true synthesis has been reported. Therefore, this compound with promising binding and complexing activity has been prepared.

## Materials and Methods

Spectrometer INOVA 300 (300 MHz, Varian Inc., Palo Alto, CA, USA) was used to measure <sup>1</sup>H NMR spectra, at their RT frequencies. Chemical shifts in (δ)–[ppm] (parts per million) were referenced to the residual signal of the solvent. Coupling

constants (J) are given in [Hz] with multiplicity: s (singlet), d (doublet), dd (doublet of the doublet), t (triplet), q (quartet), q (quintet) and m (multiplet). Tetramethylsilane was used for the calculation of <sup>1</sup>H chemical shift scales and correctly referenced using the (residual) solvent signals (2.50 and 39.52 ppm for DMSO).

All reagents and solvents were purchased from Sigma-Aldrich<sup>®</sup> (Darmstadt, Germany), Alfa-Aesar<sup>®</sup> (Ward Hill, MA, USA), Fluka<sup>®</sup> (Buchs, Switzerland) and Mikrochem<sup>®</sup> (Pezinok, Slovakia). Solvents were purified and/or dried using standard laboratory methods and stored over molecular sieves (4 Å). Column chromatography was performed using silica gel Nomasil–40–63 m (VWR<sup>®</sup>, Randor, PA, USA) and a suitable eluent according to TLC. Reaction progress was monitored by thin layer chromatography on Silufol or Alufol plates (Merck<sup>®</sup>, Darmstadt, Germany) with a UV indicator for λ = 254 nm.

Melting points (m. p.) of the prepared compounds were determined on a Boetius micro hot stage using digital thermometer TD 121 (VWR<sup>®</sup>, Randor, PA, USA) and are uncorrected.

## Experimental

### 3-Hydroxymethyl-4-oxo-1,4-dihydroquinoline 7

A sample of 0,5 g (2,3 mmol) of ethyl 1,4-dihydro-4-oxoquinoline-3-carboxylate (**4**) was added to a 100 mL flask and solved in dry THF (20 mL) equipped with a CaCl<sub>2</sub> drying stopper under argon at LT. To this solution, small portions of powdered LiAlH<sub>4</sub> (0.5 g) were added and the mixture was stirred at LT for 24 hours. Then, it was poured into a water solution of NH<sub>4</sub>Cl (50 mL, 10 %) which was afterwards filtered and the filter was washed with ethyl acetate. The filtrate was also extracted with ethyl acetate. Both parts were collected and evaporated to dryness. Yield 0.3 g (74,4 %) of 3-hydroxymethyl-4-oxo-1,4-dihydroquinoline, almost pure raw product. M. p. 183–184 °C.

Elemental analysis (calc./found): %C: 68,57/68,69, %H: 5,14/4,96 and %N: 8,00/7,69.

$^1\text{H}$  NMR ( $\text{DMSO-}d_6$ ): 4.42 d ( $\text{CH}_2$ , 2H, 4,8 Hz), 4.94 t (OH, 1H, 4,8 Hz), 7.30 t (H-8, 1H, 7,8 Hz), 7.53 d (H-6, 1H, 7,8 Hz), 7.63 t (H-7, 1H, 7,8 Hz), 7.88 bs (H-2, 1H), 8.11 d (H-5, 1H, 7,8 Hz).

## Results and Discussion

3-Hydroxymethyl derivative **7** represents an interesting tautomeric system (Scheme 1).

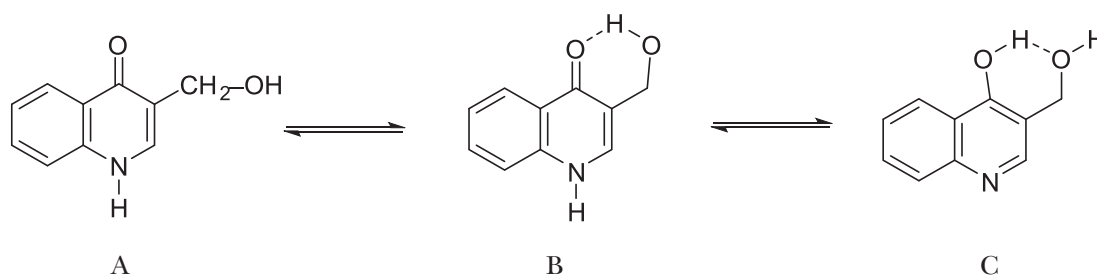
4-Oxoform B, compared to tautomer A, includes an intramolecular hydrogen bond between oxygen of the carbonyl group of the 4-pyridone system and hydrogen of the hydroxyl group. Another possible tautomer is aromatic 4-hydroxyquinoline with intramolecular hydrogen bond between hydrogen of the hydroxy group of 4-hydroxyquinoline and oxygen of the hydroxymethyl group (C) (Scheme 1).

Our first attempt was focused on the exploitation of formaldehyde as a hydroxymethylation agent considering that unsubstituted 4-quinolone (**6**, prepared from **4** via **5**, Scheme 2) reacts under the conditions of electrophilic substitution to position 3-, which has the highest electron density in the enaminone system, in agreement with previous results (Zubkov, 2003). Variation of reaction conditions (time, temperature, solvent) resulted in un-

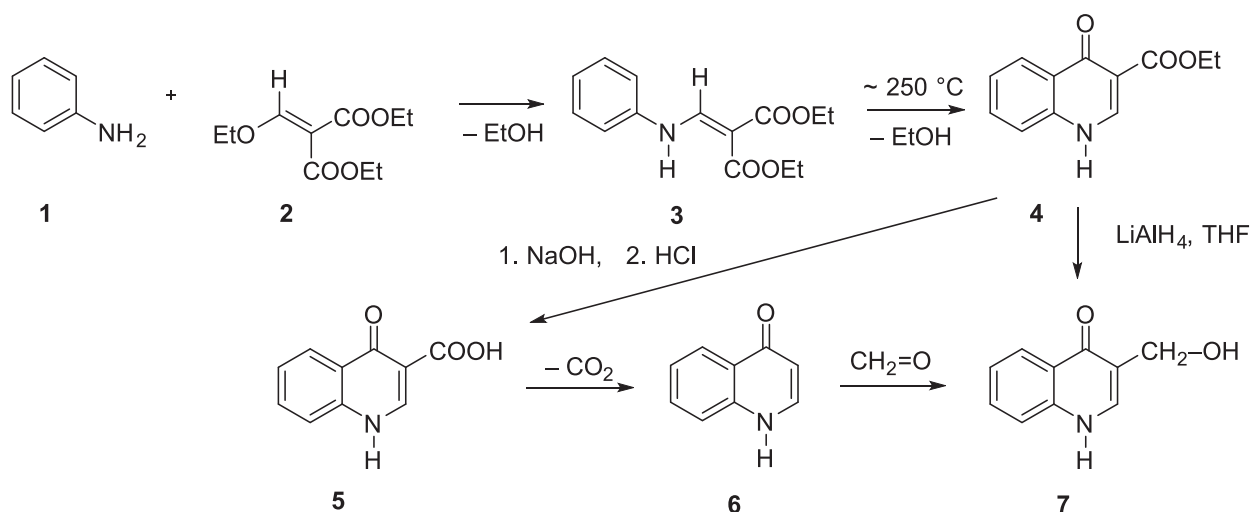
satisfactory yields, reactions produced mixtures of deeply colored and hardly separable by-products. The second strategy was based on the reduction of corresponding 3-ethoxycarbonyl derivative, easily accessible through the Gould-Jacobs reaction (Nicolson 1989). Heating of the starting equimolar mixture of aniline and diethyl ethoxymethylene malonate (EMME) to 120–130 °C for three hours (instead of 2 h at 130 °C) gave only an 83 % yield after cooling to LT (Nicolson, 1989: 96 % after cooling to -78 °C). Thermal cyclisation of diethyl anilinomethylene malonate (**3**) in diphenyl ether to reflux, including cooling of the reaction mixture, filtering off of the separated raw solid (**4**) and washing with chloroform to remove traces of **3** yielded 60 % of ethyl 4-oxo-1,4-dihydroquinoline-3-carboxylate (**4**) (Nicolson, 1989: after cyclisation in Dowtherm, an eutectic mixture of biphenyl and diphenyl ether offered a 73 % yield after recrystallisation from acetic acid). Dilution of **3** in solvent was 2,63 g/20 ml while Nicolson (1989) used the dilution of 10 g/50 ml both for 30 min.

### Acknowledgement

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**Scheme 1.** Tautomerism of 3-hydroxymethyl-1,4-dihydro-4-oxoquinoline **7**.



**Scheme 2:** Preparation of 3-hydroxymethyl-1,4-dihydro-4-oxoquinoline **7**.

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